



## COURSE DESCRIPTION CARD - SYLLABUS

Course name

Kinetics of liquids and gases

### Course

Field of study

Year/Semester

Transport

1/2

Area of study (specialization)

Profile of study

general academic

Level of study

Course offered in

First-cycle studies

polish

Form of study

Requirements

full-time

elective

### Number of hours

Lecture

Laboratory classes

Other (e.g. online)

30

Tutorials

Projects/seminars

15

**Number of credit points**

2

### Lecturers

Responsible for the course/lecturer:

Responsible for the course/lecturer:

dr Edyta Janeba-Bartoszewicz

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tel. 616652497

Faculty of Civil Engineering and Transport

### Prerequisites

Knowledge: The student knows the basics of physics and chemistry as well as the basics of thermodynamics and fluid mechanics

Skills: Student use of terminology terms in the field of mechanics, thermodynamics, physics and chemistry. Correct description of the observed phenomena, analysis of the obtained results and drawing conclusions.

Social competences. Work in an interdisciplinary team. Ability to lead a team and expand team knowledge.

### Course objective

Getting to know the basic relationships showing the physical and chemical properties of gases.



### Course-related learning outcomes

#### Knowledge

The student has extended and in-depth knowledge of physics useful for formulating and solving selected technical tasks, in particular for correct modeling of real problems

#### Skills

The student is able to properly plan and conduct perform experiments, including measurements and computer simulations, interpret the obtained results, and correctly draw conclusions

The student is able to design elements of means of transport using data on environmental protection

#### Social competences

The student is aware of the importance of knowledge in solving engineering problems, knows examples and understands the causes of malfunctioning transport systems that have led to serious financial and social losses or to serious loss of health and even life

### Methods for verifying learning outcomes and assessment criteria

Learning outcomes presented above are verified as follows:

The knowledge acquired during the lecture is verified on the basis of a written exam in the form of a test.

The skills acquired during the exercises are verified on the basis of a final test in the form of a written test.

### Programme content

Characteristics of the gas and liquid state. Thermodynamic properties: ideal, semi-perfect and real gas equations of state, compressibility factor. Viscosity of gases and liquids, depending on pressure and temperature. Influence of gases and liquids on pipeline materials. Influence of aggressive ingredients, anti-corrosion and anti-erosion protection. Combustion. Phase equilibria in multicomponent systems. Osmotic phenomena in two-component systems. Osmosis, dialysis. Donnan's membrane equilibria. Diffusion. Kinetics and mechanism of phase changes.

### Teaching methods

Information and problematic lecture with a multimedia presentation. Exercises - solving problems.

### Bibliography

#### Basic

1. J. Szargut: Termodynamika techniczna, PWN 1991
2. J. Molenda: Gaz ziemny, PWN 1999
3. H. Buchowski, W. Ufnalski „Fizykochemia gazów i cieczy”, Wydawnictwa Naukowo -Techniczne, Warszawa 2012



Additional

1. K. Pigoń, Z. Ruziewicz: Chemia fizyczna, PWN 2012

### Breakdown of average student's workload

|   | Hours | ECTS |
|---|-------|------|
| Total workload  | 60    | 2,0  |
| Classes requiring direct contact with the teacher   | 45    | 1,5  |
| Student's own work (literature studies, preparation for laboratory classes/tutorials, preparation for tests/exam, project preparation) <sup>1</sup> | 15    | 0,5  |

<sup>1</sup> delete or add other activities as appropriate